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1 A Comparative Study of Oil Palm and Japanese Beech on their  
2 Fractionation and Characterization as Treated by Supercritical  
3 Water

4  
5 Mahendra Varman and Shiro Saka\*

6  
7 Laboratory of Energy Ecosystems, Department of Socio-Environmental Energy Science,

8 Graduate School of Energy Science, Kyoto University, Yoshida-honmachi

9 Sakyo-ku, Kyoto 606-8501, Japan

10 \*Tel/Fax.: +81-(0)75-753-4738

11 \*E-mail address: saka@energy.kyoto-u.ac.jp

12  
13 **Abstract**

14 Supercritical water treatment (380°C/100MPa/8s) was applied to extractives-  
15 free oil palm trunk and sapwood portion of Japanese beech and their fractionated  
16 products were comparatively characterized, for water-soluble portion and water-  
17 insoluble portion composed of methanol-soluble portion and methanol-insoluble  
18 residue. As a result, the water-soluble portion was determined to be composed of  
19 carbohydrate-derived products such as organic acids, sugar decomposed products and  
20 lignin-derived products, etc. for both species. The oil palm, however, shows higher  
21 yield of organic acids and lignin-derived products compared to Japanese beech. The  
22 methanol-soluble portion was, on the other hand, mainly composed of lignin-derived  
23 products, and in oil palm, these lignin-derived products consisted of both syringyl and  
24 guaiacyl-type similar to Japanese beech. The methanol-insoluble residue, was also  
25 mainly composed of lignin to be 99.2% in its content in the oil palm, compared to

1 Japanese beech 93.7%. Moreover, the phenolic hydroxyl content determined by  
2 aminolysis method was higher in oil palm (36.5 PhOH/100C<sub>9</sub>) compared to Japanese  
3 beech (16.9 PhOH/100C<sub>9</sub>). Furthermore, an alkaline nitrobenzene oxidation analysis  
4 indicated that, the methanol-insoluble residue of the oil palm was less in oxidation  
5 products whereas in Japanese beech nitrobenzene oxidation products completely  
6 diminished. These lines of evidence suggest that methanol-insoluble residue is  
7 composed of lignin with more condensed-type of linkages with high phenolic  
8 hydroxyl groups. In addition, the water-soluble portion could be utilized for organic  
9 acid production, whereas the methanol-soluble portion and its insoluble residue for  
10 phenolic chemical production.

11

12 **Keywords:** supercritical water treatment; oil palm; Japanese beech; carbohydrate-  
13 derived products; lignin-derived products; organic acid production; phenolic chemical  
14 production

15

## 1 **1. Introduction**

2 Biomass shows a great potential for the production of renewable transportation  
3 fuels and chemicals. Biomass is extremely abundant resource that can be produced in  
4 agriculture, forestry and also collected as waste residues. Therefore, efficient  
5 utilization of biomass is a significant challenge for meeting the future fuel and  
6 chemical needs. For conversion of lignocellulosic biomass into fuel ethanol, however,  
7 certain challenges need to be addressed such as requirement for an efficient  
8 microorganism that is able to ferment a variety of sugars (pentoses and hexoses) [1],  
9 regulations against genetically modified organisms [2] and also challenges in  
10 producing the fuel ethanol economically. Therefore, conversion of lignocellulosic  
11 biomass into other chemicals is preferred.

12 Meanwhile, supercritical water treatment ( $>374^{\circ}\text{C}$ ,  $>22.1\text{MPa}$ ) is a clean  
13 process that is preceded without a catalyst and for a short reaction time. It has been  
14 applied to hydrolysis of cellulose [3] and wood [4, 5] for potential recovery of  
15 saccharides, dehydrated and fragmented compounds as well as organic acids and  
16 lignin-derived phenolic compounds.

17 In the present study, two biomass samples have been selected because of their  
18 anatomical differences, which are oil palm (*Elaeis guineensis*) and Japanese beech  
19 (*Fagus crenata*). Under botanical terms, oil palm is classified as one of  
20 monocotyledonous angiosperms, whereas Japanese beech, a hardwood, is classified as  
21 one of dicotyledonous angiosperms. In oil palm, vascular bundles, which consist of  
22 both xylem and phloem are scattered throughout the trunk. Whereas, in sapwood  
23 portion of hardwoods, it only consists of secondary xylem. As such, it implies that oil  
24 palm has distinct characteristics in comparison to hardwoods and therefore needs to  
25 be investigated further. With huge land coverage of oil palm especially in Indonesia,

1 Malaysia and Thailand, a large amount of the oil palm trunk is generated annually as a  
2 result of replantation after the average height of the plant exceeds 10m. It is estimated  
3 that in the year 2004 alone, 5Mton of oil palm trunk was generated in Malaysia [6].  
4 Meanwhile, Japanese beech is generated as waste wood chips. Therefore, this study  
5 will also enable us to investigate the future potential utilization of these biomass  
6 samples.

7

## 8 **2. Materials and Methods**

### 9 2.1 Sample preparation

10 The oil palm trunk was obtained from Johor Bahru, Malaysia, while sapwood  
11 portions of Japanese beech was from Kyoto, Japan. These samples were first cleaned  
12 and air-dried. The air-dried samples were then pulverized using a Wiley mill (Yoshida  
13 Seisakusho, Japan) and a Fritsch mill (Fritsch, Germany) to pass through 70-mesh  
14 sieves. The pulverized samples were then extracted with ethanol/benzene (1:2, v/v)  
15 for 8h in a Soxhlet apparatus. Subsequently these samples were oven-dried at 105°C  
16 for 10h prior to use for experiments.

17

### 18 2.2 Determination of chemical composition in biomass samples

19 Prior to the supercritical water treatment, chemical composition of those 2  
20 biomass samples were determined. Extractives-free biomass flour was used for the  
21 determination of holocellulose,  $\alpha$ -cellulose, lignin and ash contents according to refs.  
22 [7-9].

23

### 24 2.2 Supercritical water treatment and fractionation of the products

1           The supercritical water biomass conversion system used in this study was  
2 associated with a batch-type reaction vessel made of Inconel-625 with a volume of  
3 5ml [4]. The extractives-free oven-dried sample (150mg) was placed in the reaction  
4 vessel with 4.9ml distilled water of high performance liquid chromatography (HPLC)-  
5 grade (Nacalai Tesque Inc., Kyoto). Thermocouple and pressure gauge were attached  
6 to the reaction vessel. The reaction vessel was then quickly heated by immersing it  
7 into the molten tin bath preheated at 500°C. After the temperature rising time of about  
8 22s, supercritical water treatment proceeded for 8s, which represented the temperature  
9 inside the reaction vessel at an average of 380°C and pressure at 100MPa during the  
10 8s treatment. After that, the reaction vessel was immersed into water bath to quench  
11 the reaction.

12           Eventually, the supercritical water treated products was refrigerated for 12h  
13 and filtrated to separate water-soluble portion and water-insoluble residue. The  
14 obtained water-insoluble residue was then washed with 10ml HPLC-grade methanol  
15 (Nacalai Tesque Inc., Kyoto) for 10min by sonication. The methanol-insoluble  
16 residue was then separated by filtration from methanol-soluble portion as filtrates.  
17 Subsequently, the yields of these fractions were determined quantitatively after  
18 complete evaporation of methanol from methanol-insoluble residue. The yield of  
19 water-soluble portion was then estimated by subtracting the yield of methanol-soluble  
20 portion and methanol-insoluble residue from the initial amount of oven-dried sample  
21 used (150mg).

22

23   2.3 Characterization of the fractionated portions

1 Characterization of the water-soluble portion was conducted with HPLC, ion  
2 chromatography (IC), capillary electrophoresis (CE) and ultraviolet-visible (UV-Vis)  
3 spectrophotometer.

4 The HPLC analysis was carried out to quantify the presence of  
5 monosaccharides and decomposed products of saccharides using Shimadzu LC-10A  
6 under the following conditions: column; Shodex KS801, flow-rate; 1ml/min, eluent;  
7 HPLC-grade distilled water, column temperature; 80°C. The IC analysis was, on the  
8 other hand, carried out to quantify the presence of oligosaccharides and  
9 monosaccharides using Dionex ICS-3000 under these conditions: column; CarboPac  
10 PA1, column temperature; 35°C, eluent; gradient-programmed mixture of HPLC-  
11 grade distilled water, 0.2M sodium hydroxide and 1.0M sodium acetate, flow-rate;  
12 1.0ml/min.

13 Meanwhile, the CE analysis was carried out to quantify the presence of low  
14 molecular weight organic acids with an Agilent G1600A under the following  
15 conditions: column; fused-silica capillary (104cm x 75 $\mu$ m $\Phi$ ), buffer; cation pH 5.6  
16 (Agilent Co.), voltage; -30kV, temperature; 15°C, detector; UV<sub>270nm</sub> and  
17 characterization of the lignin-derived products was conducted with a Shimadzu UV-  
18 2400 UV-Vis spectrophotometer at the wavelength of 205nm.

19 Characterization of the methanol-soluble portion was conducted with Hitachi  
20 G7000M-M9000 gas chromatography-mass spectrometry (GC-MS) for qualitative  
21 analysis of low molecular weight products under these conditions: column; Varian  
22 CP-Sil 8CB (30m x 0.25mm $\Phi$ ), injector temperature; 250°C, column temperature;  
23 30°C (1min), 30°C  $\rightarrow$  250°C (5°C/min increment), 250°C (10min), carrier gas; helium,  
24 flow-rate; 1.5ml/min, emission current; 20 $\mu$ A.

1 For characterization of the methanol-insoluble residue, the determination of  
2 lignin content was conducted according to refs. [7, 10]. These results were compared  
3 with lignin content of the untreated 2 biomass samples. The phenolic hydroxyl  
4 content was determined by aminolysis method according to the procedure described  
5 by ref. [11] to determine the number of phenolic hydroxyl groups over 100  
6 phenylpropane (C<sub>9</sub>) units of lignin (PhOH/100C<sub>9</sub>); gas chromatography (Shimadzu  
7 GC14B) with flame ionization detector (FID) was used for the phenolic hydroxyl  
8 content under the following conditions: column; Shimadzu stainless steel packed with  
9 10wt% polyethyleneglycol 20M on 60/80 mesh Shincarbon A (2m x 3mm  $\Phi$ ), column  
10 temperature; 180°C (isothermal), injection port temperature; 230°C, detector  
11 temperature; 250°C, carrier gas; helium. The average weight of phenylpropane (C<sub>9</sub>)  
12 unit of lignin in oil palm was estimated by the weight of syringyl-type and guaiacyl-  
13 type C<sub>9</sub> units, considering the molar ratio of syringyl to guaiacyl moieties. For  
14 Japanese beech, it was based on hardwood data from ref. [12].

15 The alkaline nitrobenzene oxidation method employed in this study is  
16 according to ref. [13], with veratraldehyde (Nacalai Tesque Inc., Kyoto) as the  
17 internal standard. In order to quantify the nitrobenzene oxidation products, vanillin  
18 acetate, syringaldehyde acetate and *p*-hydroxybenzaldehyde acetate were prepared  
19 with the respective reagent grade chemicals (Nacalai tesque Inc., Kyoto).  
20 Quantification of nitrobenzene oxidation products was carried out by gas  
21 chromatography (Shimadzu GC14B) with FID under the following conditions:  
22 column; Shimadzu CBP-5 column (25m x 0.25mm  $\Phi$ ), column temperature; 100°C (1  
23 min), 100°C  $\rightarrow$  270°C (5°C/min increment), 270°C (10min), injection port  
24 temperature; 250°C, detector temperature; 250°C, carrier gas; helium, flow-rate;  
25 1.5ml/min.

### 3. Results and Discussion

Table 1 shows chemical composition of oil palm and Japanese beech. The cellulose content in oil palm was lower compared to Japanese beech. Hemicellulose content is higher in Japanese beech than oil palm. Meanwhile, lignin content is higher in oil palm. In contrast, oil palm also shows substantially higher ash content, which indeed help in recycling the elements from the plant to the soil if it is continuously used as mulching material.

#### 3.1 Fractionation of the products

As in Table 2, supercritical water treatment (380°C/100MPa/8s) was applied to the 2 biomass samples and the obtained yields for fractionated water-soluble portion and water-insoluble residue were investigated. In a previous work [3], a similar study has been carried out on microcrystalline cellulose under the condition of the rising time of 13s to 380°C in the molten tin bath preheated at 550°C, followed by the 380°C/100MPa treatment in the reaction vessel for 2-8s. The treatment condition is little bit different in this study due to the longer rising time in the molten tin bath (500°C), approximately 22s, to reach 380°C/100MPa in reaction vessel.

For oil palm, the yield of water-soluble portion was the highest, but similar to those in Japanese beech. It is known that the water-soluble portion consists more carbohydrate-derived products [4], which makes supercritical water treatment more efficient for decomposing the carbohydrate portion than subcritical water treatment. However, the yield of the methanol-soluble portion of the oil palm was lower than the methanol-insoluble residue. This is exactly the opposite of Japanese beech. Generally, the yield of the methanol-insoluble residue and methanol-soluble portion, are all dependent on the type of lignin linkages [4]. However, oil palms possess high ash

1 content in the Klason lignin, which could cause higher methanol-insoluble residue  
2 remaining because ash being deposited in the methanol-insoluble residue.

### 3 3.2 Characterization of water-soluble portion

4 Due to the long rising time, approximately 22s, taken for the temperature  
5 inside the reaction vessel to reach 380°C before 8s supercritical water treatment was  
6 made, neither oligosaccharides nor monosaccharides were detected; only decomposed  
7 products of saccharides such as dihydroxyacetone (DA), levoglucosan (LG), furfural  
8 (FR), organic acids and so on were detected as in Table 3 apart from lignin-derived  
9 products. The yield of unknown products which was determined by subtracting the  
10 identified products from the total is also shown.

11 The products detected in the water-soluble portion are somewhat similar  
12 between oil palm and Japanese beech. However, the yield of organic acids in Japanese  
13 beech is lower compared to that in oil palm. In oil palm, the yields of organic acids  
14 are more than 10%, whereas in Japanese beech, it is lower than 5%. Therefore, oil  
15 palm shows greater potential to be converted into organic acids by prolonging the  
16 supercritical water treatment at reduced pressure based on the study conducted with  
17 Japanese beech [5]. Organic acids are essential for subsequent conversion into  
18 methane and biodegradable polymers [5]. It could also be sold directly as a  
19 commodity chemical. Meanwhile, the yields of lignin-derived products are higher in  
20 the oil palm. It gives an indication that high non-condensed-type lignin might be  
21 present in the oil palm. This is discussed more in detail later.

22

### 23 3.3 Characterization of methanol-soluble portion

24 For the methanol-soluble portion of the 2 biomass samples, GC-MS analysis  
25 was performed. The total-ion chromatograms of the methanol-soluble portion

1 obtained by GC-MS analysis are also shown in Figure 1. Based on the GC-MS  
2 analysis, the molecular weight (MW), the mass fragmentation pattern obtained by  
3 electron ionization and the peaks identified from Figure 1 are shown in Table 4.  
4 Identification of the peaks were conducted with the retention times and mass  
5 fragmentation patterns compared with those of the authentic compounds. However,  
6 peaks Nos. 3, 4 and 5 were determined from the mass fragmentation pattern reported  
7 by ref. [14], whereas peaks Nos. 13, 16 17 and 18 were determined from ref. [15]. It  
8 could then be elucidated that these identified phenolic compounds must be mainly  
9 derived from lignin as indicated by G (guaiacyl-type) and S (syringyl-type) lignins.  
10 These compounds become insoluble as oily products when the critical state of water  
11 returns to the initial ambient state. Therefore, they are hydrophobic in nature.  
12 However, they could be fractionated mainly as the methanol-soluble portion.

13 From the previous study [4], it was found that in supercritical water, ether  
14 linkages are preferentially cleaved and smaller fraction of lignin-derived can be  
15 washed out with methanol. As a result, the residue becomes rich in condensed  
16 linkages. For the oil palm, it could be observed that the higher peaks detected were  
17 mainly from syringyl-type phenolic compounds. This is due to more syringyl-type  
18 lignin contained in oil palm, as reported previously [16]. For Japanese beech, as  
19 expected, it revealed the peaks of syringyl-type as well as guaiacyl-type phenolic  
20 compounds. This represents the nature of hardwoods. Thus, the oil palm shows  
21 similarity with hardwoods. Besides that, the methanol-soluble portion shows the  
22 potential for many phenolic compounds to be recovered, as treated by supercritical  
23 water. Phenolic compounds have great value for the production of resins, surfactants  
24 and tanning agents [17-18].

25

### 1 3.4 Characterization of the methanol-insoluble residue

2 Table 5 shows the lignin content for the methanol-insoluble residue of these 2  
3 biomass samples. It shows the lignin content for oil palm was 99.2%. For Japanese  
4 beech, however, it corresponded to 93.7%. This suggests that the methanol-insoluble  
5 residue is mostly composed of lignin and thus cellulose and hemicellulose are thought  
6 to be degraded to various compounds with low molecular weights as collected to be  
7 water-soluble portion shown in Table 3. Lignin is, thus, recovered mainly as the  
8 methanol-insoluble residue as well as methanol-soluble portion.

9 Table 5 also shows the number of the phenolic hydroxyl groups (PhOH) upon  
10 100C<sub>9</sub> units of lignin for the methanol-insoluble residue of the 2 biomass samples. It  
11 is apparent that the methanol-insoluble residue has more phenolic hydroxyl groups  
12 than the untreated biomass samples. Previously, it was demonstrated with lignin  
13 model compounds that the condensed-type linkages, such as 5-5 linkage was stable  
14 during supercritical water treatment, whereas the non-condensed-type ether linkages  
15 such as  $\beta$ -O-4 linkage was easily cleaved by supercritical water hydrolysis [4]. After  
16 the cleavage of the non-condensed-type linkages, phenolic hydroxyl groups increase.  
17 This explains the reason for the higher phenolic hydroxyl content observed in  
18 methanol-insoluble residues and it suggests that many non-condensed-type linkages  
19 are cleaved and that the residues are rich in condensed-type linkages. On the other  
20 hand, the phenolic hydroxyl content in untreated oil palm shows distinct  
21 characteristics in comparison to hardwood, whereby its content is much higher.

22 As reported previously [14], the molar ratio of syringyl to guaiacyl moieties as  
23 measured by that of syringaldehyde to vanillin (S/V) in the oil palm was 3.5 and in  
24 Japanese beech it was found to be 2.3. In addition, the yield of oxidation products of

1 the oil palm shows similarity to hardwood Japanese beech which is more than 40% as  
2 shown in Figure 2.

3 In contrast, the methanol-insoluble residue of oil palm shows small presence  
4 of nitrobenzene oxidation products whereas in Japanese beech no presence of alkaline  
5 nitrobenzene oxidation products was detected, as shown in Figure 2. These results  
6 were expected because nitrobenzene oxidation products are mainly derived from the  
7 degradation of the non-condensed-type lignin and the fact that most of these linkages  
8 are already cleaved under supercritical water treatment as mentioned above. It also  
9 suggests that methanol-insoluble residues are rich in condensed-type lignin.

10

#### 11 **4. Conclusions**

12 The characteristics of oil palm and Japanese beech after fractionation with  
13 supercritical water treatment has been presented and compared with each other. Most  
14 of the characteristics of the oil palm as one of the monocotyledonous angiosperms are  
15 similar to those of hardwood Japanese beech, one of the dicotyledonous angiosperms.  
16 However, the oil palm does show some distinct characteristics from hardwood  
17 Japanese beech such as higher ash content and higher phenolic hydroxyl content.

18 Apart from that, supercritical water technology showed the potential as the  
19 rapid and nontoxic conversion process of oil palm and Japanese beech into organic  
20 acids and the possibility for many phenolic compounds to be recovered. In addition, a  
21 good relation was found for further increasing the yield of organic acids in the oil  
22 palm by prolonging the treatment time. Although future investigations are needed for  
23 this, this study has already revealed a promising method for the conversion of biomass  
24 sources from various anatomical origins to be converted into value-added products.

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**Table 5** Lignin contents and the number of phenolic hydroxyl groups over 100 C<sub>9</sub> units of lignin (PhOH/100C<sub>9</sub>) determined by aminolysis method for the methanol-insoluble residues and untreated samples of oil palm and Japanese beech.

**Table 1** Chemical composition of the extractives-free oil palm and Japanese beech.

Sample	Chemical composition (wt%)				
	Cellulose <sup>1</sup>	Hemicellulose <sup>2</sup>	Lignin <sup>3</sup>	Ash	Others <sup>4</sup>
Oil palm	31.7	34.4	29.6	4.3	0
Japanese beech	35.6	38.0	21.8	0.6	4.0

<sup>1</sup>Cellulose =  $\alpha$ -Cellulose; <sup>2</sup>Hemicellulose = Holocellulose –  $\alpha$ -Cellulose; <sup>3</sup>Lignin = Klason lignin + acid-soluble lignin; <sup>4</sup> Others = Protein + Starch + Pectin + Losses during acid chlorite method

**Table 2** Yields of fractionated water-soluble portion and water-insoluble residue for oil palm and Japanese beech as treated by supercritical water at 380°C/100MPa/8s.

Sample	Yield (wt%)		
	Water - soluble	Water - insoluble	
		Methanol - soluble	Methanol - insoluble
Oil palm	82.1	7.5	10.4
Japanese beech	75.7	18.1	6.2

**Table 3** Yields of products in the fractionated water-soluble portion of oil palm and Japanese beech.

Sample	Water-soluble (wt%)							
	Dihydroxy-acetone	Levo-glucosan	Methyl-glyoxal	Furfural	Acetic acid	Glycolic acid	Lactic acid	Lignin-derived products
Oil palm	5.4	1.1	0.4	0.5	6.4	4.8	1.7	12.2
Japanese beech	7.6	1.4	0.2	0.2	1.4	2.5	0.8	9.8

**Table 4** Identified products in the methanol-soluble portion by its mass fragments in GC-MS analyses, as 2 biomass species treated by supercritical water (380°C/100MPa/8s).

Peak	MW	Major mass fragments	Compound	Origin	Oil palm	Japanese beech
1	124	109, 124, 81	Guaiacol	G		O
2	152	137, 152, 122	4-Ethylguaiacol	G	O	O
3	134	105, 133, 79, 134	2-Methyl-4-vinylphenol	G	O	O
4	148	133, 146, 105, 78, 92	2-Methyl-4-(1-propenyl) phenol	G		O
5	148	132, 147, 105, 78, 92	2-Methyl-4-(1-propenyl) phenol isomer	G		O
6	154	154, 139, 111	Syringol	S	O	O
7		146, 118, 117, 123, 161	unknown		O	O
8		146, 118, 117, 147, 92	unknown		O	O
9		148, 146, 92, 78, 126	unknown			O
10	164	164, 149, 131	<i>cis</i> -Isoeugenol	G	O	O
11	164	164, 149, 103, 77	<i>trans</i> -Isoeugenol	G	O	O
12	166	151, 166, 123	Acetoguaiacone	G	O	
13	182	167, 182, 168, 153	4-Ethylsyringol	S	O	O
14	180	137, 179, 122	Guaiacylacetone	G		O
15	196	167, 196, 168, 123	4-Propylsyringol	S	O	O
16	194	194, 179, 119, 91, 77	<i>cis</i> -4-Propenylsyringol	S	O	O
17	192	192, 131, 105, 90	4-Propynylsyringol	S	O	O
18	194	194, 119, 151, 179, 77	<i>trans</i> -4-Propenylsyringol	S	O	O
19	178	178, 135, 108, 77	<i>trans</i> -Coniferylaldehyde	G		O
20	196	181, 196, 153	Acetosyringone	S	O	O
21		176, 204, 177, 161, 148	unknown		O	
22	210	167, 210, 123	Syringylacetone	S	O	O
23		146, 160, 189, 206, 104	unknown		O	O
24	194	194, 167, 139, 111, 177	Ferulic acid	G		O
25		190, 204, 162, 158, 147	unknown		O	O

G: guaiacyl-type lignin; S: syringyl-type lignin; O: Compounds detected

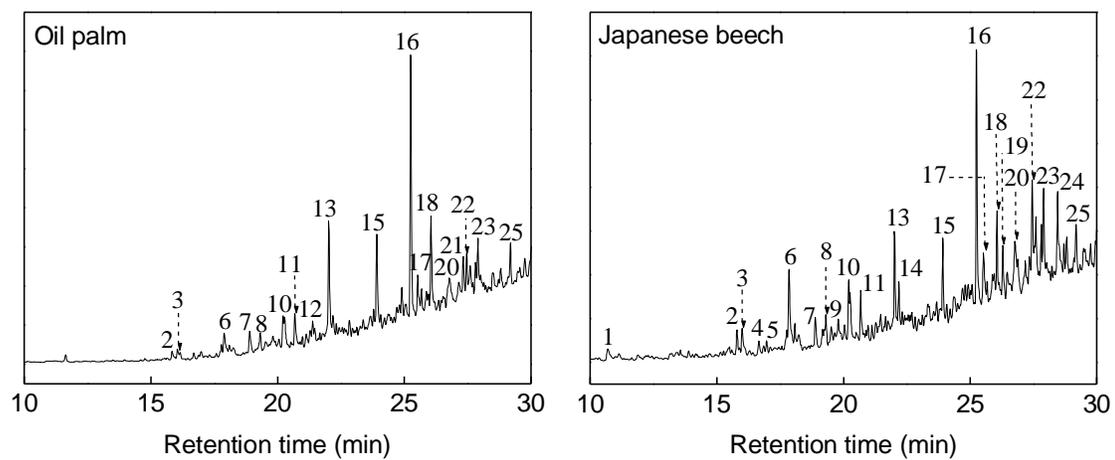
**Table 5** Lignin contents and the number of phenolic hydroxyl groups over 100 C<sub>9</sub> units of lignin (PhOH/100C<sub>9</sub>) determined by aminolysis method for the methanol-insoluble residues and untreated samples of oil palm and Japanese beech.

Sample	Methanol-insoluble residue			Untreated sample	
	Lignin (wt%)	Yield of lignin upon untreated sample (wt%)	PhOH/100C <sub>9</sub>	Lignin (wt%)	PhOH/100C <sub>9</sub>
Oil palm	99.2	10.3	36.5	29.6	28.4
Japanese beech	93.7	5.8	16.9	21.8	9.2

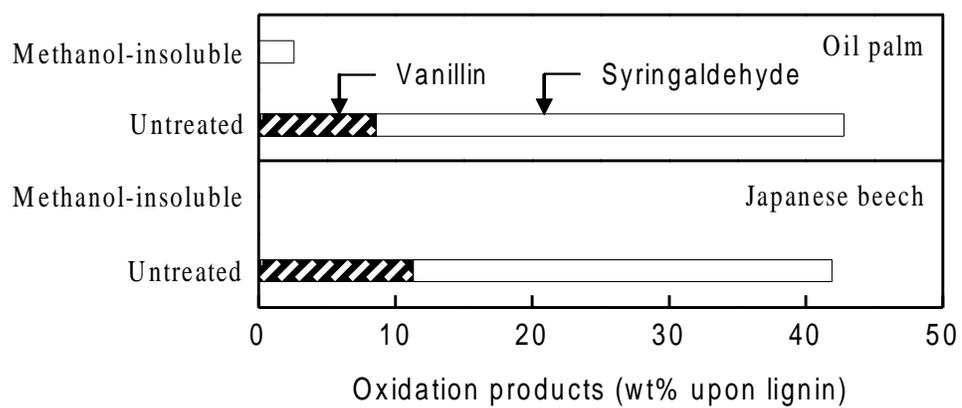
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**Figure 1** Total-ion chromatograms for the methanol-soluble portion from oil palm and Japanese beech in GC-MS analysis.

**Figure 2** Yields of the alkaline nitrobenzene oxidation products of the methanol-insoluble residues and untreated samples for oil palm and Japanese beech.



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